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**Ocr F325 June 2010  
Question**

F325 Mark Scheme  
June 2010 6 Question  
Expected Answers  
Marks Additional

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Guidance d  $\text{HNO}_3 + \text{CH}_3\text{CH}_2\text{COOH} \rightleftharpoons \text{CH}_3\text{CH}_2\text{COOH}_2^+ + \text{NO}_3^-$   
acid 1 base 2  
acid 2 base 1 2 State symbols NOT required  
ALLOW 1 AND 2 labels the other way around.  
ALLOW 'just acid' and 'base' labels throughout if linked by lines so that

**Mark Scheme for  
June 2010 - AQA,  
OCR and Edexcel**

† Write your answer to

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each question in the space provided.

Additional paper may be used if necessary but you must clearly show your Candidate Number, Centre Number and question number(s).

INFORMATION FOR CANDIDATES † The number of marks is given in brackets [ ] at the end of each question or part question.

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**CHEMISTRY A F325 -  
Science Skool!**

F325 June 2010 -

Duration: ... Energetics  
and Elements June

2015 Question 4 -

Duration: 11:26. ...

OCR A Chemistry F325  
June 2011 Q8

discussion and answer

- Duration: ...

**F325 Equilibria  
Energetics and  
Elements June 2010  
Q7**

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Chemistry. ... OCR Unit  
5 F325 June 2013 Past  
Paper Work through ...  
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and Analysis June 2010  
Q4-5 - Duration: 10:14.  
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10:14. ...

### **F325 June 2010**

F325 Mark Scheme  
June 2010 7 Question  
Expected Answers  
Marks Additional  
Guidance 4 a i  
Complete circuit (with  
voltmeter) and salt



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bridge linking two half-cells □ Pt electrode in solution of  $\text{Fe}^{2+}/\text{Fe}^{3+}$  □ Ag in solution of  $\text{Ag}^{+}$  □  
3 DO NOT ALLOW 'solution of a silver halide', e.g. AgCl (as these are insoluble) but

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**OCR Unit 5 (F325)  
Chemistry Revision -**

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Tutor**

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2010 question - The  
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Unit 5 OCR Chemistry  
A-level; June 2010 QP -  
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A-level; June 2011 MS  
...

**OCR Unit 5  
Chemistry Past  
Papers - Physics &  
Maths Tutor**

F325 Mark Scheme  
June 2011 1 Question  
Answer Mark Guidance  
1 (a) (The enthalpy  
change that  
accompanies) the  
formation of one mole

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Question Paper

of a(n ionic) compound  
from its gaseous ions  
(under standard  
conditions) 2 IGNORE  
'Energy needed' OR  
'energy required'  
ALLOW as alternative  
for compound: lattice,  
crystal, substance,  
solid Note: 1st mark  
requires 1 mole

**F325 Mark Scheme**  
**June 2011 Question**  
**Answer Mark**  
**Guidance 1 ...**

OCR (Oxford  
*Page 14/26*)

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Cambridge and RSA) is a leading UK awarding body, providing a wide range of qualifications to meet the needs of candidates of all ages and abilities. OCR qualifications ... F325 Mark Scheme June 2012 Question Answer Marks Guidance 2 (a) (iii) FIRST, CHECK THE ANSWER ON ANSWER LINE . IF.

**Advanced GCE Unit  
F325: Equilibria,**  
*Page 15/26*

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**Energetics and  
Elements**

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F325 Mark Scheme  
January 2012 6

Question Expected  
answers Marks

Additional guidance 2 a  
(The enthalpy change  
that accompanies) the  
formation of one mole  
of a(n ionic) compound  
from its gaseous ions

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(under standard conditions) 2 IGNORE 'Energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid, product

**Advanced GCE Unit  
F325: Equilibria,  
Energetics and  
Elements**

F325 Mark Scheme  
June 2010 8 Question  
Expected Answers  
Marks Additional

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Question Paper

Guidance c A fuel cell converts energy from reaction of a fuel with oxygen into a voltage/electrical energy □ 5

ANNOTATIONS MUST BE USED

ALLOW combustion for reaction of fuel with oxygen/reactants

ALLOW a fuel cell requires constant supply of fuel

**3 ALLOW F 1450 B  
736 G E 76 D -642**

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**FIVE FOUR THREE b  
ALLOW ...**

F325 Mark Scheme

June 2011 1 Question

Answer Mark Guidance

1 (a) (The enthalpy  
change that

accompanies) the

formation of one mole

of a(n ionic) compound

from its gaseous ions

(under standard

conditions) 2 IGNORE

'Energy needed' OR

'energy required'

ALLOW as alternative

for compound: lattice,

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Question Paper

crystal, substance,  
solid Note: 1st mark  
requires 1 mole

**Advanced GCE Unit  
F325: Equilibria,  
Energetics and  
Elements**

F325 Mark Scheme  
June 2014 3 Question  
Answer Marks

Guidance 1 (a) (i)

$2\text{K}^+(\text{g}) + \text{S}^-(\text{g}) + \text{e}^-$   
 $2\text{K}^+(\text{g}) + \text{S}^{2-}(\text{g})$   $2\text{K}(\text{g})$   
 $+ \text{S}(\text{g})$  3 Mark each  
marking point

independently Correct

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species AND state  
symbols required for  
each mark For S<sup>2-</sup>, DO  
NOT ALLOW S<sup>-2</sup> For e<sup>-</sup>,  
ALLOW e For e<sup>-</sup> only,  
IGNORE any state  
symbols added ALLOW  
k and s It can be very  
difficult distinguishing  
K from k; S from s

**Oxford Cambridge  
and RSA - A-Level  
Chemistry**

an effective person?  
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Chemistry A . Mark  
Scheme for January  
2011 . OCR (Oxford  
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