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## **Naming Compounds Worksheet With Answers**

Answers - Naming Chemical Compounds

1) NaBr sodium bromide 2)  $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$

2) 2 calcium acetate 3)  $\text{P}_2\text{O}_5$

diphosphorus pentoxide 4)  $\text{Ti}(\text{SO}_4)_2$

titanium(IV) sulfate 5)  $\text{FePO}_4$  iron(III)

phosphate 6)  $\text{K}_3\text{N}$  potassium nitride 7)

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SO<sub>2</sub> sulfur dioxide 8) CuOH copper(I)  
hydroxide 9) Zn(NO<sub>2</sub>)<sub>2</sub> zinc nitrite 10)  
V<sub>2</sub>S<sub>3</sub> vanadium(III) sulfide

## **Naming Ionic Compounds Practice Worksheet**

Answers - Naming Chemical Compounds  
. Name the following chemical  
compounds: 1) NaBr sodium bromide. 2)

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Ca(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> calcium acetate. 3) P<sub>2</sub>O<sub>5</sub> diphosphorus pentoxide. 4) Ti(SO<sub>4</sub>)<sub>2</sub> titanium(IV) sulfate. 5) FePO<sub>4</sub> iron(III) phosphate. 6) K<sub>3</sub>N potassium nitride. 7) SO<sub>2</sub> sulfur dioxide. 8) CuOH copper(I) hydroxide. 9) Zn(NO<sub>2</sub>)<sub>2</sub> zinc nitrite. 10) V<sub>2</sub>S<sub>5</sub>

## Answers - Naming Chemical

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## **Compounds**

Naming Compounds Worksheet 1. The formulas and common names for several substances are given below. Give the systematic names for these substances.

Common Name Chemical Formula

Systematic Name a. Sugar of lead  $\text{Pb}(\text{CH}_3\text{COO})_2$  Note:  $\text{CH}_3\text{COO}$  is a polyatomic ion. See your polyatomic ion table for

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name and charge. b. Blue vitrol  $\text{CuSO}_4$   
c. Epsom salts ...

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making publication describes, as well as  
to ...

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Solutions for the Naming Ionic  
Compounds Practice Worksheet 1)  
ammonium chloride 2) iron (III) nitrate 3)

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titanium (III) bromide 4) copper (I)  
phosphide 5) tin (IV) selenide 6) gallium  
arsenide 7) lead (IV) sulfate 8) beryllium  
bicarbonate 9) manganese (III) sulfite  
10) aluminum cyanide 11) Cr(PO<sub>4</sub>)<sub>3</sub>

## **Naming Compounds Practice Worksheet**

Mixed Ionic Covalent Compound Naming

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Naming Compounds What's it Made of?

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Metal and non-metal 2 non-metals 3 or more elements polyatomic compound covalent compound ionic compound USE GREEK PREFIXES Put prefixes in front of element names to tell how many atoms are there. Don't use "mono" for first name, but always for second name. Li 2S  
Metal and non-metal— ionic

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- Answers Name these compounds. They may be either ionic or covalent. 1)  $\text{LiOH}$  lithium hydroxide 2)  $\text{PBr}_3$  phosphorus tribromide 3)  $\text{Na}_2\text{SO}_4$  sodium sulfate 4)  $(\text{NH}_4)_2\text{S}$  ammonium sulfide 5)  $\text{CaCO}_3$  calcium carbonate 6)  $\text{CF}_4$  carbon tetrafluoride 7)  $\text{NaNO}_3$  sodium nitrate 8)  $\text{P}_2\text{S}_3$  diphosphorus trisulfide 9)  $\text{Al}(\text{NO}_3)_3$  aluminum nitrate 10)  $\text{Mg}(\text{OH})_2$



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magnesium hydroxide Write the formulas for the following compounds.

## **Naming Ionic Compounds - Answer Key**

Mixed Naming Worksheet #2 W 303  
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Program Write the names of the

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following chemical compounds: ... Write the names of the following chemical compounds: 1)  $\text{Fe}(\text{CN})_3$  iron (III) cyanide 2)  $\text{K}_4\text{C}$  potassium carbide 3)  $\text{AuF}_3$  gold (III) fluoride 4)  $\text{N}_2\text{O}$  dinitrogen monoxide 5)  $\text{Ag}_3\text{N}$  ...

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Naming Chemical Compounds - Answers.

Name the following ionic compounds: 1)

NaBr sodium bromide. 2) CaO calcium

oxide. 3) Li<sub>2</sub>S lithium sulfide. 4) MgBr<sub>2</sub>

magnesium bromide. 5) Be(OH)<sub>2</sub>

beryllium hydroxide Write the formulas

for the following ionic compounds: 6)

potassium iodide KI. 7) magnesium

oxide MgO. 8) aluminum chloride AlCl<sub>3</sub>

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Naming Organic Compounds Practice  
EXERCISES A. Identify the class of the  
following compounds. For any alkanes,  
alkenes, alkynes, aromatic compounds,  
carboxylic acids or alcohols, provide the  
IUPAC name of the molecule. For the

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four special monosubstituted benzenes, use the common name. 1) I 8) Br CH<sub>3</sub> 2) CH<sub>3</sub> H<sub>2</sub>C CH<sub>3</sub> H<sub>3</sub>C CH CH<sub>2</sub>CH<sub>2</sub> 9) H<sub>3</sub>C H<sub>3</sub>C C ...

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Covalent Compound Naming Worksheet  
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Compounds Solutions. Write the formulas for the following covalent compounds: 1) antimony tribromide  $\text{SbBr}_3$ . 2) hexaboron silicide  $\text{B}_6\text{Si}$  3) chlorine dioxide  $\text{ClO}_2$ . 4) hydrogen iodide  $\text{HI}$  5) iodine pentafluoride  $\text{IF}_5$ . 6) dinitrogen trioxide  $\text{N}_2\text{O}_3$ .

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